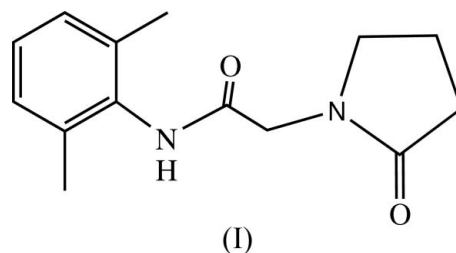


Pin-Liang Wang,* Hai-Bo Wang,
Wei-Lin Ding, Zhi-Qian Liu and
Zhi-Tao XingDepartment of Applied Chemistry, College of
Science, Nanjing University of Technology,
Xinmofan Road No. 5 Nanjing, Nanjing
210009, People's Republic of ChinaCorrespondence e-mail:
wanghaibo@njut.edu.cn

Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.046
 wR factor = 0.121
Data-to-parameter ratio = 9.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.***N*-(2,6-Dimethylphenyl)-2-(2-oxo-1-pyrrolidin-1-yl)-
acetamide**The title compound, $\text{C}_{14}\text{H}_{18}\text{N}_2\text{O}_2$, a derivative of pyrrolidone, was synthesized by *N*-acylation and *N*-alkylation reactions, with 2,6-xylylidine as the starting material. In the crystal structure, there is an intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond.Received 12 July 2006
Accepted 9 August 2006

Comment

2-Pyrrolidones represent an important class of five-membered heterocycles. Pyrrolidone derivatives are of great interest because of their biological properties. Some derivatives of pyrrolidone improve cerebral functions (Kamihara *et al.*, 1995). Some show high efficacy as antidementia agents (Kisofukushima *et al.*, 1990). The title compound, (I), is one of the efficient drugs for the treatment of Alzheimers disease (Rene *et al.*, 2001). The molecular structure of (I) is shown in Fig. 1. In the crystal structure, there is an intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond (Table 1).

Experimental

2,6-Xylylidine (20 mmol) was dissolved in toluene (300 ml), and sodium carbonate (15 mmol) was dissolved in water (100 ml). The two solutions were combined and chloroacetyl chloride (15 mmol) was added in a nitrogen stream at room temperature, followed by stirring for 2 h. 2-Pyrrolidone (30 mmol) and sodium amide (8 mmol) were then added slowly dropwise. After stirring the mixture at 333–343 K for 5 h, hot water (50 ml) was added and the mixture was allowed to cool with stirring. The crystals precipitated in the aqueous layer were recovered by filtration and dried under reduced pressure to obtain the title compound. The pure compound was obtained by recrystallization from ethyl acetate. Crystals suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution.

Crystal data

| | |
|--|-----------------------------------|
| $\text{C}_{14}\text{H}_{18}\text{N}_2\text{O}_2$ | $Z = 4$ |
| $M_r = 246.30$ | $D_x = 1.220$ Mg m $^{-3}$ |
| Orthorhombic, $P2_12_12_1$ | Mo $K\alpha$ radiation |
| $a = 6.3390$ (13) Å | $\mu = 0.08$ mm $^{-1}$ |
| $b = 9.3130$ (19) Å | $T = 293$ (2) K |
| $c = 22.712$ (5) Å | Block, white |
| $V = 1340.8$ (5) Å 3 | $0.30 \times 0.20 \times 0.20$ mm |

Data collection

Enraf–Nonius CAD-4
diffractometer
 $\omega/2\theta$ scans
Absorption correction: ψ scan
(North *et al.*, 1968)
 $T_{\min} = 0.966$, $T_{\max} = 0.989$
3018 measured reflections

1545 independent reflections
1114 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.050$
 $\theta_{\max} = 26.0^\circ$
3 standard reflections
every 200 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.121$
 $S = 1.02$
1545 reflections
170 parameters
H atoms treated by a mixture of
independent and constrained
refinement

$w = 1/[\sigma^2(F_o^2) + (0.0622P)^2 + 0.07P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.15 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.15 \text{ e } \text{\AA}^{-3}$
Extinction correction: *SHELXL97*
Extinction coefficient: 0.025 (4)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------|----------|-------------|-------------|---------------|
| $N1-H1\cdots O1^i$ | 0.80 (3) | 2.07 (3) | 2.846 (3) | 165 (3) |

Symmetry code: (i) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$.

H atoms bonded to C atoms were positioned geometrically at distances of 0.93–0.96 \AA and included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl C})$. The methyl groups were allowed to rotate but not to tip. The amide H atom was refined freely. In the absence of anomalous scatterers, Friedel pairs were merged.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

References

Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.

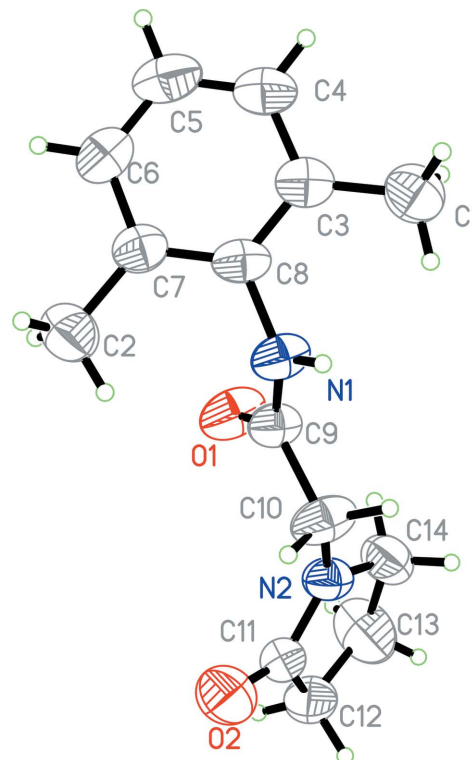


Figure 1

A view of the molecular structure of the title compound. Displacement ellipsoids are drawn at the 30% probability.

- Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.
Kamihara, S., Kaneuchi, T., Uchiyama, K. & Terada, T. (1995). US Patent No. 5 461 157.
Kisofukushima, S. I., Matsumoto, Y. K., Nagasaki, H. K. & Okaya, Y. A. (1990). US Patent No. 4 933 354.
North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
Rene, E., Lixin, Q., Alan, K. & Lianyun, Z. (2001). WO Patent No. 0 183 449.
Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
Siemens (1996). *SHELXTL*. Version 5.06. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.